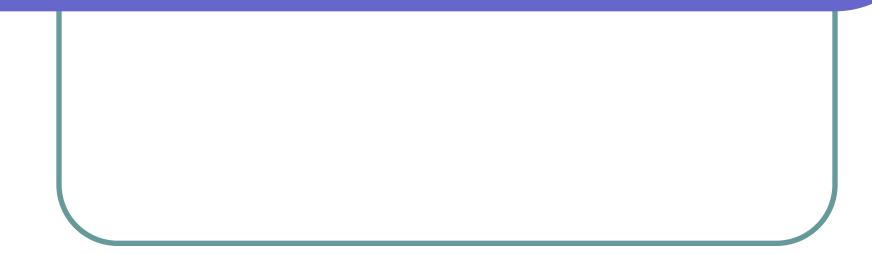
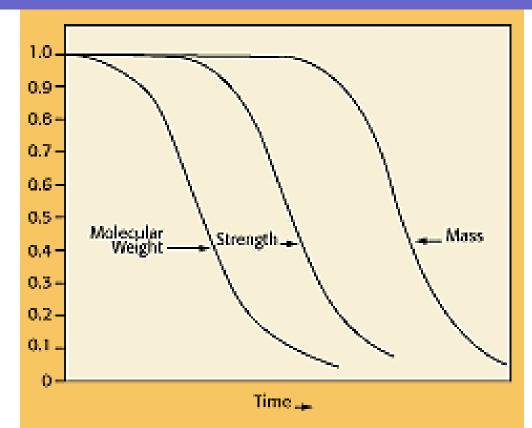
Biodegradable Polymers: Chemistry, Degradation and Applications Lecture 7 : Prof.Dr.Jenan S.Kashan

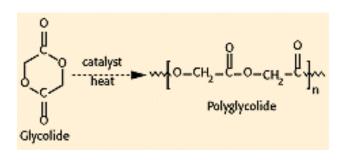


What is Polymer Degradation?



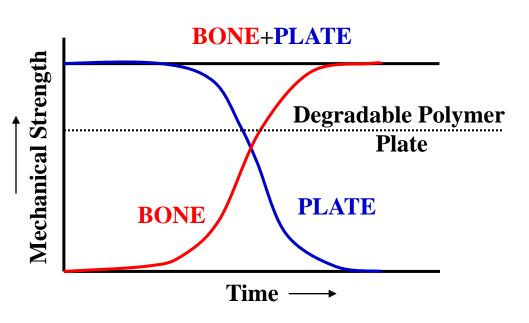
polymers were synthesized from glycolic acid in 1920s

At that time, polymer degradation was viewed negatively as a process where properties and performance deteriorated with time.

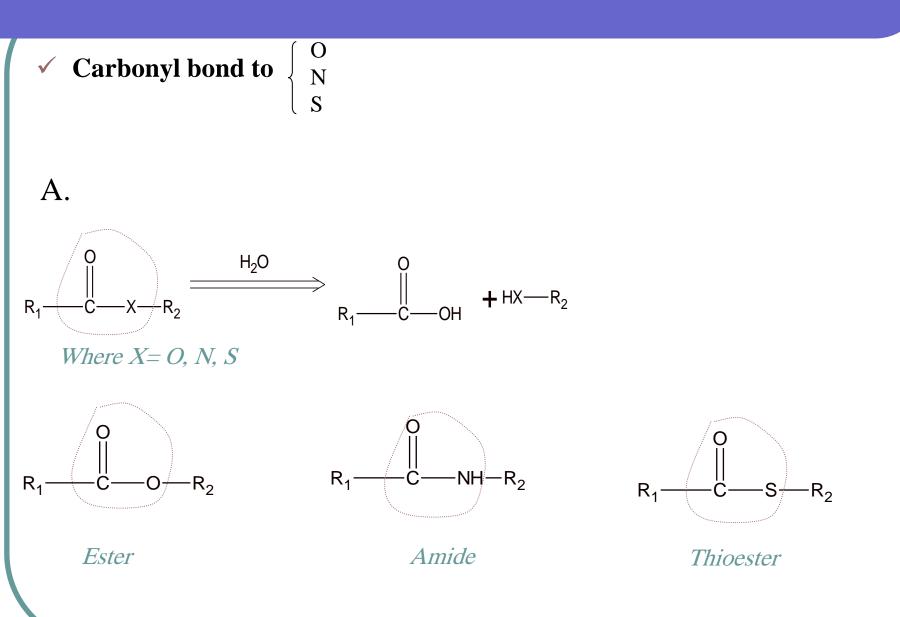


Why Would a Medical Practitioner Like a Material to Degrade in the Body?

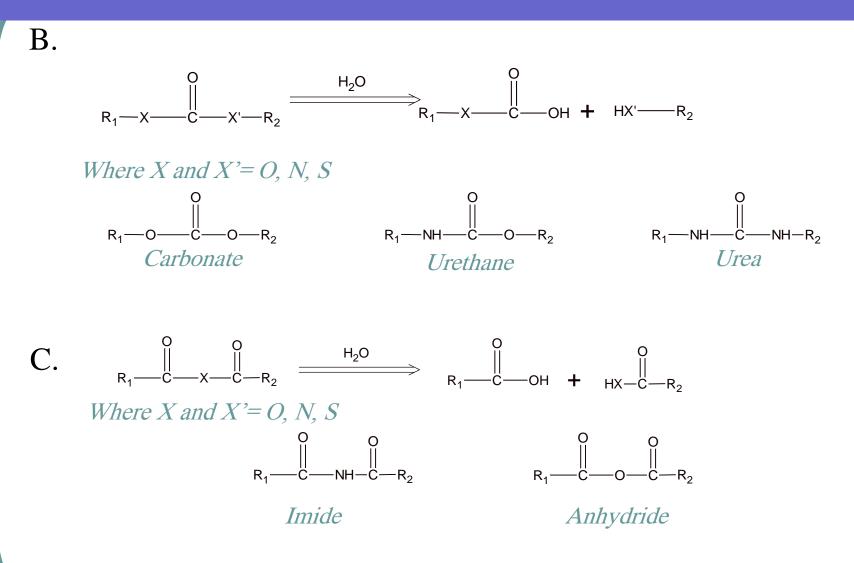
- Do not require a second surgery for removal
- Avoid stress shielding
- Offer tremendous potential as the basis for controlled drug delivery



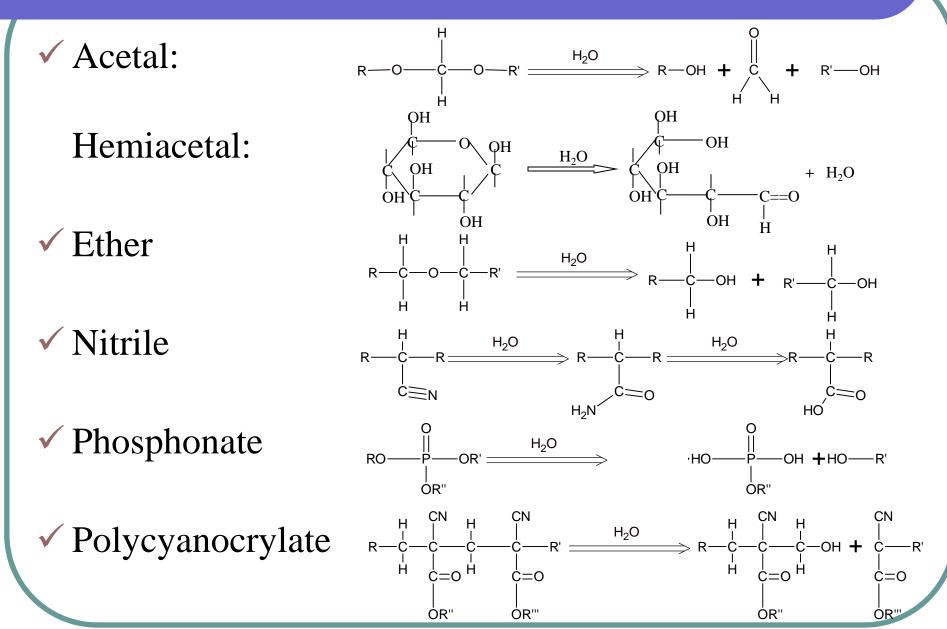
Biodegradable Polymers



Biodegradable Polymers



Biodegradable Polymers



Biodegradable Polymers Used for Medical Applications

- Natural polymers
 - Fibrin
 - Collagen
 - Chitosan
 - Gelatin
 - Hyaluronan ...
- Synthetic polymers
 - PLA, PGA, PLGA, PCL, Polyorthoesters ...
 - Poly(dioxanone)
 - Poly(anhydrides)
 - Poly(trimethylene carbonate)
 - Polyphosphazenes ...

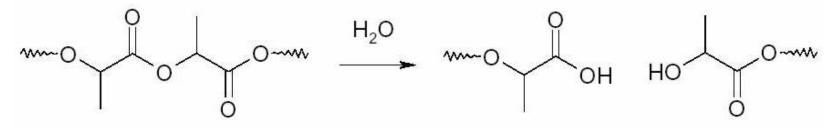
Synthetic or Natural Biodegradable Polymers Why Do We Prefer Synthetic Ones?

- Tailor-able properties
- Predictable lot-to-lot uniformity
- Free from concerns of immunogenicity
- Reliable source of raw materials

Degradation Mechanisms

- Enzymatic degradation
- Hydrolysis

(depend on main chain structure: anhydride > ester >



- Homogenous degradation
- Heterogenous degradation

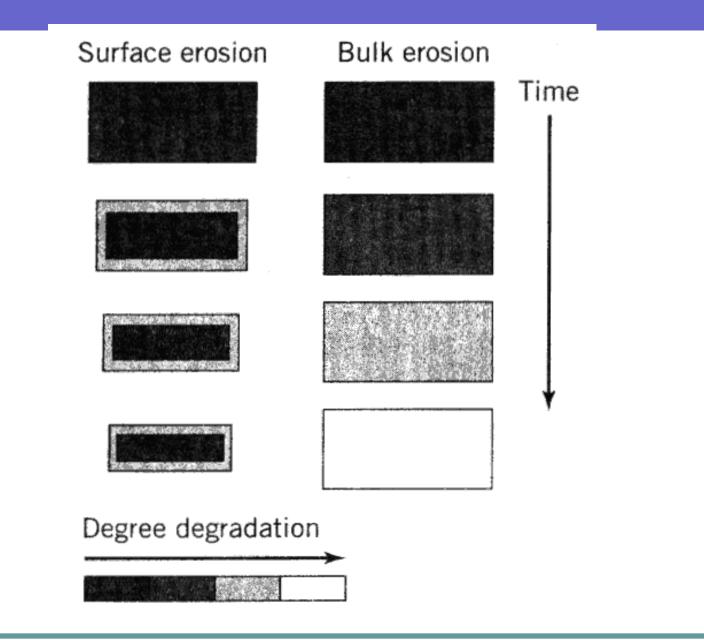
Degradation can be divided into 4 steps:

- water sorption
- reduction of mechanical properties (modulus & strength)
- reduction of molar mass
- weight loss

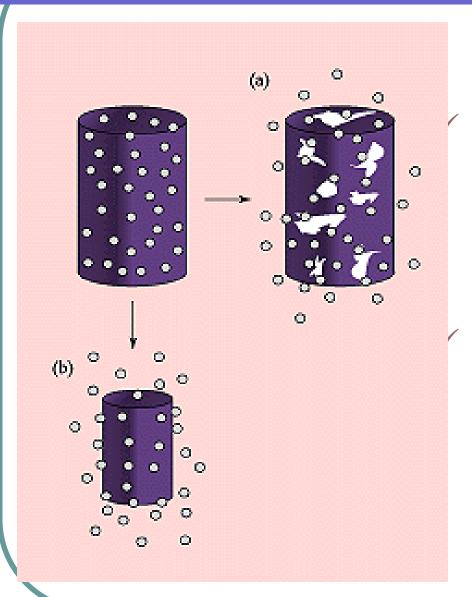
Degradation Schemes

- Surface erosion (poly(ortho)esters and polyanhydrides)
 - Sample is eroded from the surface
 - Mass loss is faster than the ingress of water into the bulk
- Bulk degradation (PLA,PGA,PLGA, PCL)
 - Degradation takes place throughout the whole of the sample
 - Ingress of water is faster than the rate of degradation

Polymer Degradation by Erosion (1)



Erodible Matrices or Micro/Nanospheres



(a) ✓ Bulk-eroding system

(b)Surface-eroding system

Molding (formation of drug matrix)

- compression molding
- melt molding
- solvent casting

Molding (compression molding) (1)

- Polymer and drug particles are milled to a particle size range of 90 to 150 µm
- Drug / Polymer mix is compressed at ~30,000 psi
- Formation of some types of tablet / matrix

Molding (melt molding / casting) (1)

- Polymer is heated to ~10°C above it melting point (T_m) to form a viscous liquid
- Mix drug into the polymer melt
- Shaped by injection molding

Molding (melt molding / casting) (2)

Advantages

- More uniform distribution of drug in polymer
- Wide range of shapes possible

Disadvantages

- Thermal instability of drugs (heat inactivation)
- Drug / polymer interaction at high temperature
- Cost

Molding (Solvent casting) (1)

- Co-dissolve drug and polymer in an organic solvent
- Pour the drug / polymer solution into a mold chilled under dry ice
- Allow solvent to evaporate
- Formation of a drug-polymer matrix

Molding (Solvent casting) (2)

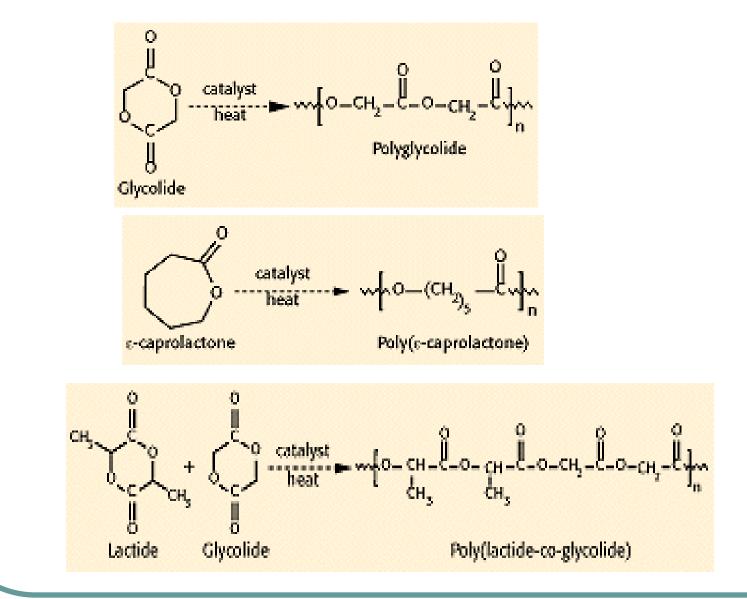
Advantages

- Simplicity
- Room temperature operation
- Suitable for heat sensitive drugs

Disadvantages

- Possible non-uniform drug distribution
- Proper solvents for drugs and polymers
- Fragility of the system
- Unwanted matrix porosity
- Use of organic solvents / Solvent residues

Polyesters





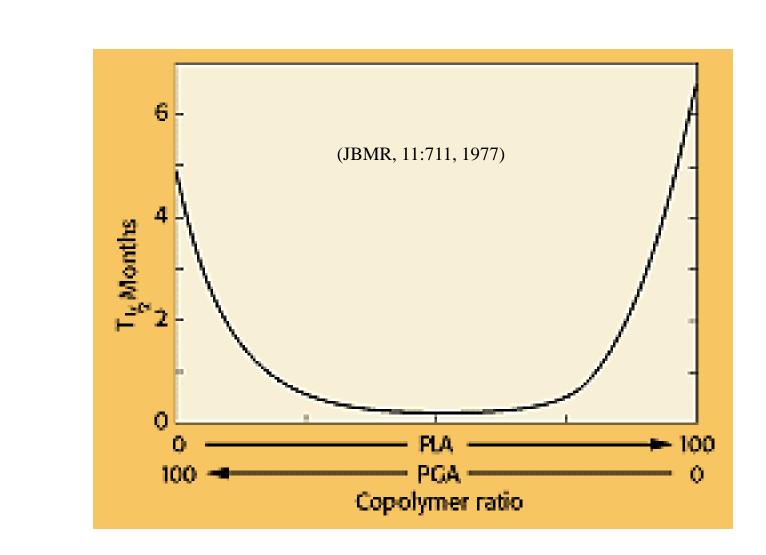
Properties	PLA	PS	PVC	PP
Yield Strength, MPa	49	49	35	35
Elongation, %	2.5	2.5	3.0	10
Tensile Modulus, GPa	3.2	3.4	2.6	1.4
Flexural Strength, MPa	70	80	90	49

Mobley, D. P. Plastics from Microbes. 1994

Factors Influence the Degradation Behavior

- Chemical Structure and Chemical Composition
- Distribution of Repeat Units in Multimers
- Molecular Weight
- Polydispersity
- Presence of Low Mw Compounds (monomer, oligomers, solvents, plasticizers, etc)
- Presence of Ionic Groups
- Presence of Chain Defects
- Presence of Unexpected Units
- Configurational Structure
- Morphology (crystallinity, presence of microstructure, orientation and residue stress)
- Processing methods & Conditions
- Method of Sterilization
- Annealing
- Storage History
- Site of Implantation
- Absorbed Compounds
- Physiochemical Factors (shape, size)
- Mechanism of Hydrolysis (enzymes vs water)

Poly(lactide-co-glycolide) (PLGA)



Factors That Accelerate Polymer Degradation

- More hydrophilic backbone.
- More hydrophilic endgroups.
- More reactive hydrolytic groups in the backbone.
- Less crystallinity.
- More porosity.
- Smaller device size.

Methods of Studying Polymer Degradation

- Morphological changes (swelling, deformation, bubbling, disappearance...)
- Weight lose
- Thermal behavior changes
 - Differential Scanning Calorimetry (DSC)
- Molecular weight changes
 - Dilute solution viscosity
 - Size exclusion chromatograpgy(SEC)
 - Gel permeation chromatography(GPC)
 - MALDI mass spectroscopy
- Change in chemistry
 - Infared spectroscopy (IR)
 - Nuclear Magnetic Resonance Spectroscopy (NMR)
 - TOF-SIMS

Medical Applications of Biodegradable Polymers

- Wound management
 - Sutures
 - Staples
 - Clips
 - Adhesives
 - Surgical meshes
 - Orthopedic devices
 - Pins
 - Rods
 - Screws
 - Tacks
 - Ligaments



- Dental applications
 - Guided tissue regeneration Membrane
 - Void filler following tooth extraction
- Cardiovascular applications
 - Stents
 - Intestinal applications
 - Anastomosis rings
- Drug delivery system
- Tissue engineering